

by B. Crossland and J. D. Williams



Explosive welding, which is a solid-phase welding process, was probably discovered by chance by Philipchuk,^{1*} who states that he first observed the effect when explosively forming an aluminium U-channel on a steel die. It was found that the U-channel could not be removed from the die because a circular-shaped area had become welded to the die. From more recent work it would appear that an excessive explosive charge must have been used in this forming exercise to have achieved the impact velocity required for welding. Allen, Mapes, and Wilson² carried out experiments concerned with the impact of right cylindrical bullets fired at thin targets and they noted a rippling of the surface of the bullet. Abrahamson,³ who was concerned with the rippling or wave action, continued the work of Allen *et al.* and observed, when firing a steel bullet against an oblique copper target, that adhesion occurred between the steel and the copper; he shows a photomicrograph of the wavy interface typical of explosive welding. This, of course, leads to the conclusion that explosive welding is associated with the oblique impact between the two surfaces to be welded. Crossland and Bahrani⁴ also noted that it was well known during the First World War that a bullet or shrapnel could stick to metal surfaces which they impacted, though it was not appreciated that this could be the basis of a welding process.

It is necessary to briefly define welding to get a correct perspective of explosive welding in relation to other welding processes. Simply, welding is a process for

joining two or more solid components by local coalescence or union across an interface. The essential conditions for any form of welding are that the two surfaces before welding should be absolutely clean and uncontaminated and that these surfaces should be brought into contact. It is impossible to produce such surfaces by normal mechanical or chemical cleaning processes but under extremely high vacuum conditions Bowden and Tabor⁵ and Keller⁶ have produced nearly perfectly clean surfaces. If such surfaces are brought into contact, adhesion will occur between the asperities. This adhesion is greatly improved if, in addition to the normal force, a tangential force is applied that is not sufficient to cause sliding.

If two perfectly clean and atomically flat surfaces of the same metal are brought together, interatomic repulsive and attractive forces will come into play and equilibrium will be reached at a particular interatomic distance when the potential energy of the system is a minimum. The strength of the bond will be influenced by factors such as crystallographic misorientation across the interface, and diffusion and recrystallisation, which are very much dependent on temperature. The situation in relation to adhesion of dissimilar metals, which may have not only different atomic spacing but also a different structure, is obviously much more complex but nevertheless adhesion can still occur as a consequence of the interatomic forces.

Various clearly recognisable and distinguishable welding processes have been developed, but all of them are basically processes for removing the contaminant surface layers to allow adhesion to occur between clean metal surfaces. Four distinct welding processes can be recognised. First, there is fusion welding, in which the surfaces of the two metals are melted by the application of heat and the contaminant surface films are brought to the surface of the melt pool or go into solution. Many sources of the heat required for fusion welding have been developed and also methods of reducing the oxidation in the region of the weld. Secondly, there is flow welding, in which a third low-melting-

point metal is used together with a suitable flux to wet the surfaces of the solid base metals of the two components to be joined, as for example in brazing, silver soldering, &c. Thirdly, pressure welding, either hot or cold, has been widely practised. In this process the two surfaces to be joined are compressed or hammered together with such force as to cause plastic flow at the interface, with an associated increase in the area of contact. This surface distortion breaks up the contaminant surface film and creates virgin surfaces where adhesion can take place. The effectiveness of the process is greatly improved if surface sliding also occurs during the plastic flow. Finally, we have explosive welding, in which the surface of one of the members is effectively peeled off to form a high-velocity metal jet which scours the surface of the other component. The two clean metal surfaces produced are then pressed together by the explosive pressure.

There are obvious limitations with some of these welding processes. For example, lead with a boiling point of 1620°C (1893 K) could hardly be fusion welded to steel, with a melting point of 1500°C (1773 K). Indeed, it is impossible to weld metals with vastly different melting points by fusion welding. There are, of course, formidable metallurgical problems in fusion welding of some metals and these problems are even more complex when welding together dissimilar metals. Correspondingly, it is unlikely that lead could be pressure welded to steel, as the lead would plastically deform much more readily than the steel and consequently no virgin surface could be produced on the latter. It is apparent that materials of basically different plastic properties cannot be pressure welded together. With explosive welding, melting is not a necessary condition, nor do the plastic properties of the two metals being welded impose any limitations on the process. It is, for example, possible to explosively weld a soft, low-melting-point metal such as lead to a hard, high-melting-point metal.

The main limitation with explosive welding is that, so far, it has proved possible to weld together only very simple

* Recently our attention has been drawn to an earlier report of welding having occurred under impacting conditions generated by explosives (L. R. Carl, *Metal Progress*, 1944, 46, (7), 102).

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